

## STIMULATORY EFFECT OF PLATINUM(IV) ION ON THE PRODUCTION OF SUPEROXIDE RADICAL FROM XANTHINE OXIDASE AND MACROPHAGES

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**Abstract**—Superoxide radical ( $\text{O}_2^-$ ) production was measured spectrophotometrically using NADH and lactate dehydrogenase (LDH) in a xanthine oxidase(XOD) plus hypoxanthine(HX) system and in an isolated guinea pig macrophages system. Sodium platinum(IV) chloride ( $\text{Na}_2\text{PtCl}_6$ :  $2.5 \times 10^{-4}$ – $1 \times 10^{-3}$  M) enhanced the production of  $\text{O}_2^-$  in both systems (2–10 times). The degree of the enhancement was dependent on incubation time, basal level of  $\text{O}_2^-$  production and concentration of  $\text{Na}_2\text{PtCl}_6$ . The stimulated  $\text{O}_2^-$  production in the XOD system was inhibited by luminol (*O*-aminophthalhydrazide) and that in the macrophages was inhibited by an anti-inflammatory drug, Diclofenac sodium (Dc). These results show that platinum (IV) ion is either a potent stabilizer of  $\text{O}_2^-$  or a stimulator of  $\text{O}_2^-$  production as are paraquat or streptonigrin. This specific character of platinum (IV) ion may explain its bactericidal and inflammation-inducing properties.

Many metal ions such as  $\text{Mn}^{+2}$ ,  $\text{Hg}^{+2}$ ,  $\text{Cu}^{+2}$ ,  $\text{Fe}^{+2}$ ,  $\text{Fe}^{+3}$ ,  $\text{Ni}^{+2}$ ,  $\text{Co}^{+2}$  in concentrations of  $10^{-6}$ – $10^{-3}$  M inhibit the production of superoxide radical ( $\text{O}_2^-$ ) from isolated guinea-pig macrophages [1]. Other metal ions such as  $\text{Ca}^{+2}$ ,  $\text{Ba}^{+2}$ ,  $\text{Cd}^{+2}$  and  $\text{Pb}^{+2}$  inhibit  $\text{O}_2^-$  production to a far less extent at a concentration of  $10^{-3}$  M.  $\text{Pt}^{+4}$  ( $\text{Na}_2\text{PtCl}_6$ ) even stimulates  $\text{O}_2^-$  production.  $\text{O}_2^-$  from granulocytes is known to be bactericidal [2,3], and certain platinum compounds have been reported to inhibit cell division of gram-negative bacteria [4]. Excess production of  $\text{O}_2^-$  may be cause of inflammation. The prostaglandin phase of rat carrageenan foot oedema is completely inhibited by intravenous injections of superoxide dismutase (SOD: 0.5–2.0 mg/kg) which specifically breaks down  $\text{O}_2^-$  [5]. Platinum complexes are also reported to induce human atopic hypersensitivity [6]. Many *in vivo* effects of the platinum compounds seem to be explained by the ability of  $\text{Pt}^{+4}$  ions to stimulate  $\text{O}_2^-$  production.

### MATERIALS AND METHODS

The absorbance at 340 nm of NADH (0.96  $\mu\text{mole}$ ) was continuously recorded with a Shimadzu Multipurpose MPS-5000 spectrophotometer at 37°. Superoxide dismutase (SOD: Sigma's product from bovine blood), which is known to react specifically with  $\text{O}_2^-$ , completely inhibited the NADH oxidation induced by 0.1 U/ml xanthine oxidase (XOD: from buttermilk) plus 80  $\mu\text{M}$  hypoxanthine (HX) or by macrophage suspensions. Reaction mixtures in this experiment contained 5  $\mu\text{M}$  lactate dehydrogenase (LDH: from rabbit muscle) and either 8 mM Veronal-acetate-HCl buffer pH 6.5 or 125 mM  $\text{NaH}_2\text{Na}_2\text{HPO}_4$  buffer pH 6.5. In some experiments, 25 mM phosphate buffer was also used. Phosphate anion is essential for the production of  $\text{O}_2^-$  from macrophages and the optimum pH for its production is 6.5. The details were reported in the preceding report [1]. The

method used in the present experiment to determine the amount of  $\text{O}_2^-$  is a modification of that of Chan *et al.* [7], who introduced the LDH and NADH method.

Small amounts of  $\text{O}_2^-$  were produced even in media containing no XOD or macrophages, and this production was influenced by various agents. Stimulatory effects of agents on  $\text{O}_2^-$  production were, therefore, calculated by the following formula:

Stimulation (%)

$$= \frac{\Delta A(\text{agent} + \text{S}) - \Delta A(\text{agent only})}{\Delta A(\text{S only}) - \Delta A(\text{medium only})}$$

where  $\Delta A$  is the difference in absorption at 340 nm before and after the reaction, and S is the source of  $\text{O}_2^-$  i.e. XOD + HX or macrophage suspension. The pH of all the agents used was adjusted to 6.5 before addition. Diclofenac sodium (Dc.) and luminol solutions contained up to 0.1% of *N,N'*-dimethylformamide (DMF) to effect solution: this amount had little effect on  $\text{O}_2^-$  production.

### RESULTS

The degree of NADH oxidation was linear up to 10 min with 0.1 U/ml XOD plus 80  $\mu\text{M}$  HX (Fig. 1). This oxidation was more than doubled by addition of  $\text{Na}_2\text{PtCl}_6$  ( $5 \times 10^{-4}$  M) and was nearly completely inhibited with SOD (14 U/ml = 4  $\mu\text{g/ml}$ ). Without XOD,  $\text{Na}_2\text{PtCl}_6$  had no stimulatory effect on basal  $\text{O}_2^-$  production. At higher  $\text{Na}_2\text{PtCl}_6$  concentrations, the interval of linear oxidation was shortened. Moreover, at  $2.5 \times 10^{-3}$  M the platinum became inhibitory at 5 min; a lower  $\text{Na}_2\text{PtCl}_6$  concentration of  $2.5 \times 10^{-4}$  M required a time lag of at least 5 min to stimulate  $\text{O}_2^-$  production.

The stimulation of NADH oxidation was not due to the stimulation of LDH activity by  $\text{Na}_2\text{PtCl}_6$ . The LDH activity was measured in terms of the decrease

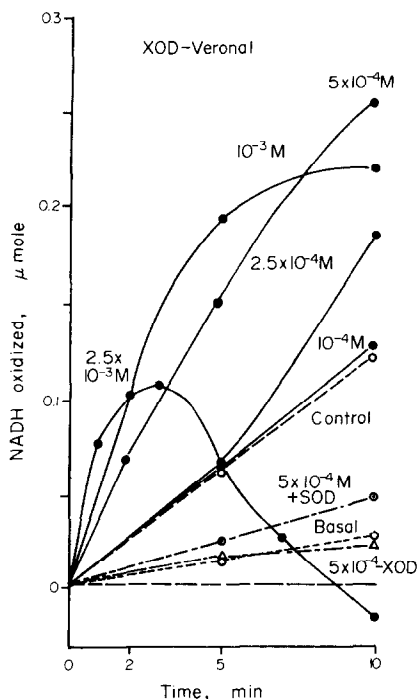


Fig. 1. Time-course of  $\text{'O}_2^-$  production effected by 0.1 U/ml XOD plus 80  $\mu\text{M}$  HX in various concentrations of  $\text{Na}_2\text{PtCl}_6$  in 8 mM Veronal-acetate-HCl buffer pH 6.5. The final concentration of SOD added was 14.0 U/ml (= 4  $\mu\text{g}$  protein/ml).

in NADH absorption at 340 nm in 30 sec incubation with 0.1 mM sodium pyruvate and  $8 \times 10^{-3}$  U/ml LDH at 37°.  $\text{Na}_2\text{PtCl}_6$  ( $10^{-4}$  M– $10^{-3}$  M) never stimulated the LDH activities. In a concurrently conducted experiment, sodium oxamate (a specific LDH inhibi-

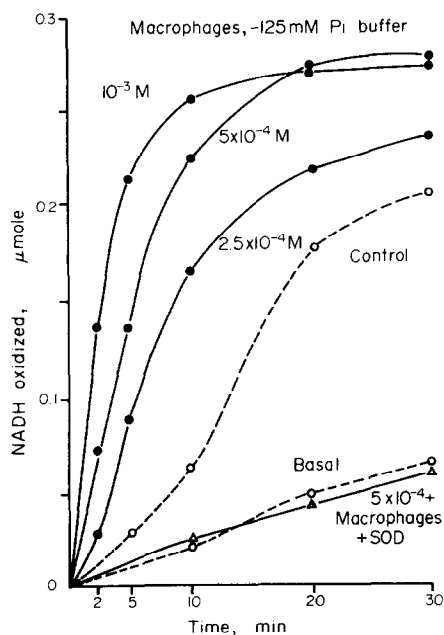


Fig. 2. Time-course of  $\text{'O}_2^-$  production effected by  $2.6 \times 10^6$  macrophages/ml in various concentrations of  $\text{Na}_2\text{PtCl}_6$  in 125 mM  $\text{NaH}_2\text{-Na}_2\text{HPO}_4$  buffer pH 6.5. The final concentration of SOD added was 14.0 U/ml.

tor) inhibited the LDH activity by 94 per cent at a concentration of  $10^{-3}$  M and 74 percent at  $10^{-4}$  M.

The  $\text{'O}_2^-$  production by the isolated guinea-pig macrophages was also stimulated by  $\text{Na}_2\text{PtCl}_6$  (Fig. 2). These findings suggest that the platinum salt stabilizes  $\text{'O}_2^-$  or facilitates  $\text{'O}_2^-$  production. The platinum salt may activate the XOD system or enzyme(s) that

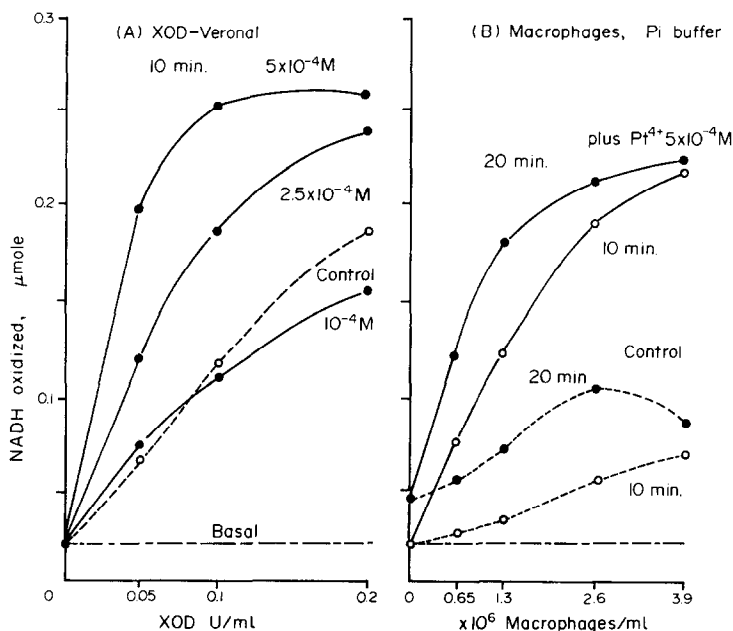


Fig. 3. Stimulation of  $\text{'O}_2^-$  production by  $\text{Na}_2\text{PtCl}_6$  in different  $\text{'O}_2^-$  producing conditions. Reaction mixture contains buffer solutions, LDH, macrophage suspension or XOD + HX and  $\text{Pt}^{4+}$  (—) or no  $\text{Pt}^{4+}$  (---) solutions. Reaction was started by the addition of NADH and incubated for 10 or 20 min. (A) 0.1 U/ml XOD + 80  $\mu\text{M}$  HX in 8 mM Veronal-acetate-HCl buffer pH 6.5. (B) Macrophages in 125 mM  $\text{NaH}_2\text{-Na}_2\text{HPO}_4$  buffer pH 6.5.

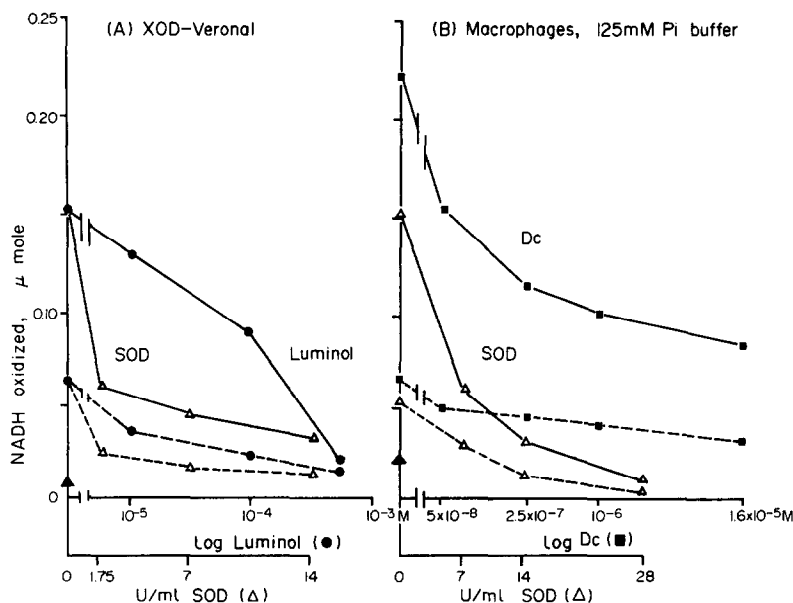


Fig. 4. Inhibition of control and  $\text{Pt}^{4+}$ -stimulated  $\text{'O}_2^-$  productions by SOD ( $\Delta$ ), luminol ( $\bullet$ ) and Diclofenac sodium (Dc:  $\blacksquare$ ). Control (---) and with  $5 \times 10^{-4}$  M  $\text{Na}_2\text{PtCl}_6$  (—). (A) 0.1 U/ml XOD + 80  $\mu\text{M}$  HX in 8 mM Veronal-acetate-HCl buffer pH 6.5 was incubated for 5 min. (B)  $1.8 \times 10^6$  macrophages/ml for SOD test or  $2.6 \times 10^6$  macrophages/ml for Dc test in 125 mM  $\text{NaH}_2\text{-Na}_2\text{HPO}_4$  buffer pH 6.5 was incubated for 10 min. ( $\blacktriangle$ ) NADH oxidation by  $5 \times 10^{-4}$  M  $\text{Na}_2\text{PtCl}_6$  without XOD + HX or macrophages.

produce  $\text{'O}_2^-$  in macrophages, but the exact mechanism of enhanced  $\text{'O}_2^-$  production by  $\text{Pt}^{4+}$  is not clear. Fig. 3 shows the differences in  $\text{'O}_2^-$  stimulation under various  $\text{'O}_2^-$  producing conditions. In both the XOD and macrophage systems, the stimulation by  $\text{Na}_2\text{PtCl}_6$  is greater in conditions where small amounts of  $\text{'O}_2^-$  are being produced. Factors that govern the stimulatory effect comprise incubation time, concentration of  $\text{Na}_2\text{PtCl}_6$  and the amount of  $\text{'O}_2^-$  production. Phosphate anion concentration is important in the case of macrophages [1].

Fig. 4 shows that the stimulation of NADH oxidation by  $\text{Na}_2\text{PtCl}_6$  is due to  $\text{'O}_2^-$  production and not to other oxidants, because SOD inhibited the  $\text{Pt}^{4+}$ -stimulated NADH oxidation. The inhibition was dose-dependent just as in the control NADH oxidation by the XOD system and by the macrophage suspension. Luminol, a  $\text{'O}_2^-$  scavenger [8], inhibited NADH oxidation in the XOD system. Diclofenac sodium (Dc), a non-steroidal anti-inflammatory drug and a potent inhibitor of  $\text{'O}_2^-$  production in the macrophage system (not in the XOD system), demon-

Table 1. Stimulation of  $\text{'O}_2^-$  production effected by  $\text{Na}_2\text{PtCl}_6$

'O <sub>2</sub> <sup>-</sup> generating source	Buffer	Time (min)	Control net NADH oxidation (μmole)	Stimulation ratio (Na <sub>2</sub> PtCl <sub>6</sub> conc. M)		
				2.5 × 10 <sup>-4</sup>	5 × 10 <sup>-4</sup>	1 × 10 <sup>-3</sup>
XOD (U/ml)						
0.05	V	5	0.020	7.2	10.7	—
0.10	V	5	0.058	1.1	2.1	2.4
0.20	V	5	0.092	2.0	2.4	—
0.05	P(b)	10	0.078	1.0	1.9	3.0
Macrophages (× 10 <sup>6</sup> /ml)						
2.6	P(a)	10	0.011	4.0	6.6	9.3
2.6	P(a)	20	0.045	3.0	3.2	3.8
2.6	P(b)	5	0.026	3.7	4.8	7.9
2.2	P(b)	5	0.018	3.2	6.6	—
2.6	P(b)	10	0.041	3.6	5.1	—
2.8	P(b)	10	0.042	2.8	5.1	—
2.2	P(b)	10	0.034	2.4	5.4	—

The stimulatory ratio was calculated as described in the text. Control net NADH oxidation was set as 1.0. V, 8 mM Veronal-acetate-HCl buffer pH 6.5; P(a), 25 mM  $\text{NaH}_2\text{-Na}_2\text{HPO}_4$  buffer pH 6.5; P(b), 125 mM  $\text{NaH}_2\text{-Na}_2\text{HPO}_4$  buffer pH 6.5.

strated dose-dependent inhibition of the NADH oxidation which was stimulated by  $5 \times 10^{-4}$  M  $\text{Na}_2\text{PtCl}_6$ .  $\text{Pt}^{+4}$  can perhaps be applied for screening anti-inflammatory drugs with economy of macrophages [1].

#### DISCUSSION

To my knowledge only a small number of agents have been reported to stimulate  $\text{O}_2^-$  production. Curnutte *et al.* [9] showed that fluoride (20 mM) increased  $\text{O}_2^-$  production about 5 times at 10 min and 2 times at 20 min when measured by the absorbance of cytochrome *c* reduced by human granulocytes. This stimulation was verified in our study using guinea-pig macrophages and a LDH + NADH assay system, but was not as marked as in their study, and disappeared after 20 min Cytochalasin B (5  $\mu\text{g}/\text{ml}$ ), contrary to the results of Curnutte's study, was shown to inhibit  $\text{O}_2^-$  production by macrophages [1]. This discrepancy may be due to differences in the assay method. Cytochrome *c* might have the wrong configuration to allow this enzyme to trap quickly  $\text{O}_2^-$  generated on the cell membrane [10, 11]. Cytochalasin E was stimulatory when examined by the nitroblue tetrazolium assay and guinea-pig leucocytes [10]. Paraquat (a herbicide) damaged human lungs [12, 13] and was reported to enhance  $\text{O}_2^-$  production *in vitro* [14]. Streptonigrin (an antibiotic, 1.0  $\mu\text{g}/\text{ml}$ ) was non-bactericidal but when combined with a small amount of  $\text{O}_2^-$ , regenerated enough  $\text{O}_2^-$  to become bactericidal [15].

$\text{Na}_2\text{PtCl}_6$  seems to fall in the category of these kinds of stimulators of  $\text{O}_2^-$  production or to be an efficient  $\text{O}_2^-$  stabilizer. This platinum salt is easily available from commercial sources, and is hopefully applicable for detecting traces of  $\text{O}_2^-$  generated from various materials. Furthermore, the platinum salt makes possible the use of smaller amounts of macrophages for screening anti-inflammatory agents [1].

The inflammation-inducing capacity of  $\text{Na}_2\text{PtCl}_6$ , was also studied (unpublished result). A saline solution of  $\text{Na}_2\text{PtCl}_6$  (4  $\mu\text{moles}/\text{guinea-pig paw}$ ) caused acute lethal effect, but when injected as 10% water in 90% vegetable oil suspension, it caused chronic paw swelling and profound damage of joint connective tissues and of bones from 3 days to more than 2 weeks. Equimolar amounts of copper acetate, ferric chloride, lead chloride and cadmium chloride caused no such swelling or damage.

$(\text{NH}_3)_2\text{PtCl}_4$ ,  $(\text{NH}_3)_2\text{PtCl}_6$  and their derivatives, which are known as anti-tumor agents [16], were not

available so their effects on  $\text{O}_2^-$  production could not be studied. These complex salts are reported to destabilize the DNA-helix *in vitro* [17]. Other anti-tumor agents, Bleomycin [18] and Mitomycin C [19] are also supposed to attack tumor cells by the formation of  $\text{O}_2^-$ . There is no report that  $\text{Na}_2\text{PtCl}_6$  itself has an anti-tumor effect, but it is worth considering the relationship between the  $\text{O}_2^-$  production and the anti-tumor effect of platinum compounds.  $(\text{NH}_3)_2\text{PtCl}_4$  is reported to accumulate in plasma and in the organs [20].

Platinum compounds also depress the skin allograft reaction in mice [21] and inhibit production of antibody to sheep erythrocytes in mice spleen cells [22]. The stimulatory effect of  $\text{Pt}^{+4}$  ions on  $\text{O}_2^-$  production may explain the effect observed in *in vivo* studies.

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